## List of Practical's Chemistry 9th

**Chapter I** 

**Experiment 1** Separate the given mixture of iron fillings and sand by Physical method.

## **Chapter 5**

raperiment 2 Determine the melting point of given compounds (Naphthalene and Biphenyl).

**Experiments 3** Determine the boiling point of given liquids (Acceton and Ethyl alcohol) and Toluen.

**Experiment 4** Demonstrate sublimation using ammonium chloride.

**Experiment 5** Separate naphthalene from given mixture of sand and naphthalene by sublimation.

Chapter No. 06

**Experiment 6** Prepare 100cm<sup>3</sup> of 0.1 M sodium hydroxide solution from the given 1M solution.

**Experiment 7** Prepare 100cm<sup>3</sup> of 0.01M Na<sub>2</sub>CO<sub>3</sub> solution from the given 0.1M solution.

**Experiment 8** Prepare 100cm<sup>3</sup> of 0.01 M hydrochloric acid solution from the given 0.1 M solution.

**Experiment 9** Prepare pure copper Sulphate Crystals from the given Impure sulphate.

Experiment 10 Demonstrate that temperature affects the solubility

Chapter 07

Experiment 11 Demonstrate the conductivity of different given solution.

Chapter 08

**Experiment 12** Demonstrate that some processes absorb energy.

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